

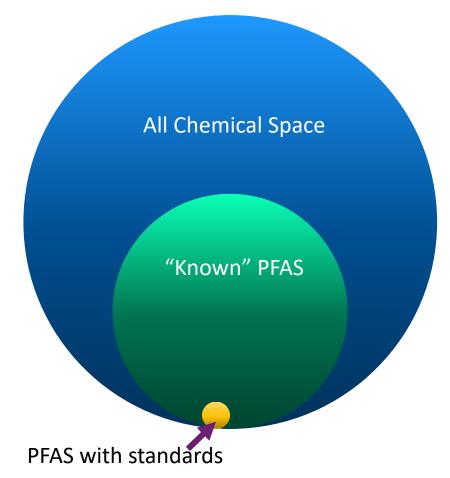
Non-Targeted and Suspect Identification of Per- and Polyfluoroalkyl Substances through Databases and Other Tools

**Benjamin Place, Research Chemist National Institute of Standards & Technology** 



## Per- and Polyfluoroalkyl Substances

- Estimates for the total number of different PFAS in the environment and/or used in varying industries can be over 4,000.
- There are a limited number of analytical standards available for these compounds.
  - Estimated ~100 standards commercially available.
  - This makes traditional targeted approaches (LC-MS/MS) more difficult
- Non-Targeted Analysis and Suspect Screening Analysis are two approaches towards the detection of known and unknown PFAS without the availability of authentic analytical standards.





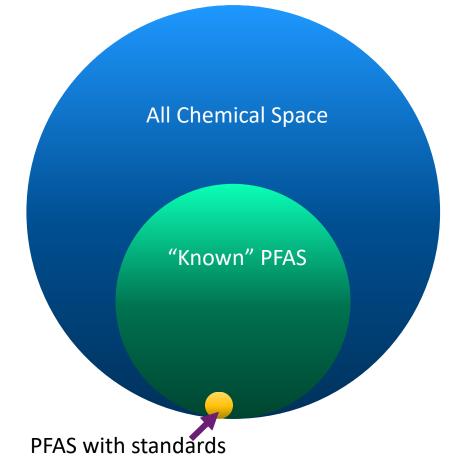


## Non-Targeted Analysis – An Overview

Non-targeted analysis (NTA) is a theoretical concept that can be broadly defined as the characterization of the chemical composition of any given sample without the use of *a priori* knowledge regarding the sample's chemical content.

Suspect screening analysis (SSA) is the identification of chemicals and/or chemical classes detected by an instrument, typically a mass spectrometer, by comparison to a predefined user list or library containing known chemicals of interest.









## Non-Targeted Analysis – Analytical Workflow

Sample Preparation

Chromatography

Mass Spectrometry

Data Analysis

- Storage/handling
- Extraction
- Clean-Up

- ChromatographyType (GC vs. LC)
- Column Phase
- Mobile Phases

- High resolution mass spectrometry (typically)
- Full Scan MS1
- Data Dependent
   (DDA) or Data
   Independent (DIA)
   MS2

More on next slide





Peak Picking Formula Assignment Fragmentation Interpretation Interpretation





Algorithm to detect peaks for all relevant\* features

Determination of elemental formula from measured m/z and isotopic pattern matching

Fragmentation interpretation (user expertise or *in silico*)

Weight of evidence leads to possible compound identification

NTA

**Peak Picking** 

Formula Assignment

Fragmentation Interpretation

Identification

Extract peaks for all possible suspects

Cross-listing measured m/z with predicted m/z of suspects

Mass Spectral Libraries (empirical or *in silico*)

Ranking of most probable matches within suspect list





Algorithm to detect peaks for all relevant\* features

Determination of elemental formula from measured *in/z* and isotopic pattern matching

Fragmentation interpretation (user expertise or in silico)

Weight of evidence leads to possible compound identification

NTA

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Fragmentation interpretation (user expenses on inserting silico)

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NTA

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Extract peaks for all possible suspects

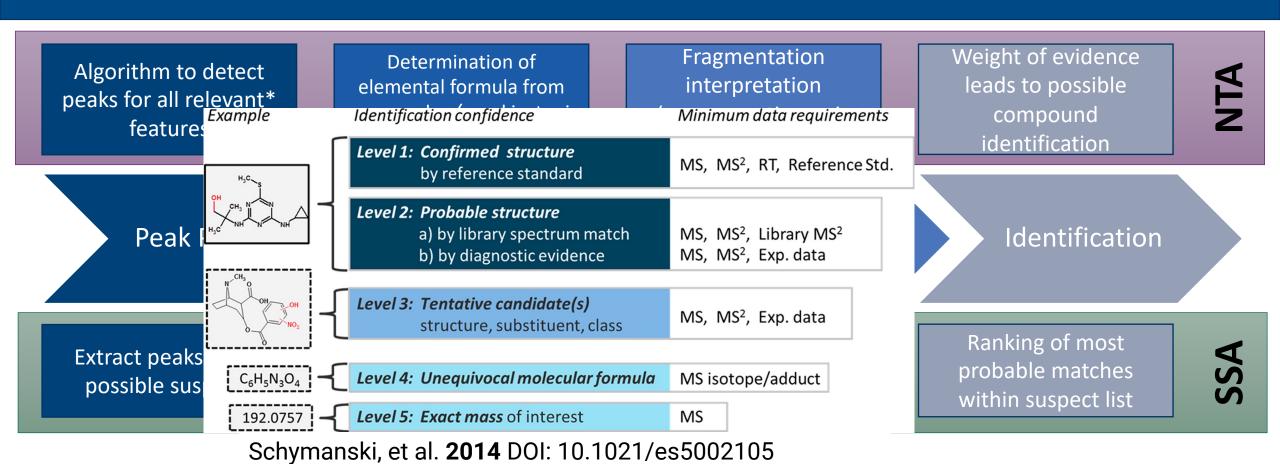
Cross-listing measured m/z with predicted m/z of suspects

Mass Spectral Libraries (empirical or *in silico*)

Ranking of most probable matches within suspect list











There are specific properties of PFAS that can be used to selectively screen the data.





# There are specific properties of PFAS that can be used to selectively screen the data.

➤ Low to negative mass defect



Article pubs.acs.org/est

Discovery of 40 Classes of Per- and Polyfluoroalkyl Substances in Historical Aqueous Film-Forming Foams (AFFFs) and AFFF-Impacted Groundwater

Krista A. Barzen-Hanson, <sup>†</sup> Simon C. Roberts, <sup>∇,‡</sup> Sarah Choyke, <sup>§</sup> Karl Oetjen, <sup>‡</sup> Alan McAlees, <sup>∥</sup> Nicole Riddell, <sup>∥</sup> Robert McCrindle, <sup>⊥</sup> P. Lee Ferguson, <sup>§</sup> Christopher P. Higgins, <sup>\*,‡</sup> and Jennifer A. Field\*, <sup>#</sup>

<sup>1</sup>H = 1.0078 amu <sup>19</sup>F = 18.9984 amu

 $C_8H_{15}O_2 = 143.1072 \text{ amu}$ (mass defect = 0.1072 amu)  $C_8F_{15}O_2 = 412.9658 \text{ amu}$ (mass defect = -0.0342 amu)



Article pubs.acs.org/es

Identification of Novel Perfluoroalkyl Ether Carboxylic Acids (PFECAs) and Sulfonic Acids (PFESAs) in Natural Waters Using Accurate Mass Time-of-Flight Mass Spectrometry (TOFMS)

Mark Strynar,\*\*,† Sonia Dagnino,†,‡ Rebecca McMahen,†,‡ Shuang Liang,†,‡ Andrew Lindstrom,† Erik Andersen,† Larry McMillan, § Michael Thurman, Imma Ferrer, I and Carol Ball





100- Buckeye AFFF - Positive Ionization Mode

200

There are specific properties of PFAS that can be used to selectively screen the data. %<sup>60∃</sup>

- ➤ Low to negative mass defect
- ➤ Co-occurrence of homologs



pubs.acs.org/est

80=

20-

m/z 100

300

pubs.acs.org/est

800

Place and Field (2012)

\* FAB Matrix

DOI: 10.1021/es301465n

♦ x:1:2 fluorotelomer betaine

900

∆ x:3 fluorotelomer betaine

Identification of Novel Fluorochemicals in Aqueous Film-Forming Foams Used by the US Military

Benjamin J. Place<sup>†</sup> and Jennifer A. Field\*,<sup>‡</sup>

Identification of Novel Fluorinated Surfactants in Aqueous Film Forming Foams and Commercial Surfactant Concentrates

514

500

414

400

632

700

614

600

Lisa A. D'Agostino and Scott A. Mabury\*





# There are specific properties of PFAS that can be used to selectively screen the data.

- ➤ Low to negative mass defect
- Co-occurrence of homologs
- Specific fragmentation patterns related to common classes of PFAS



Article pubs.acs.org/est

Discovery of 40 Classes of Per- and Polyfluoroalkyl Substances in Historical Aqueous Film-Forming Foams (AFFFs) and AFFF-Impacted Groundwater

Krista A. Barzen-Hanson, <sup>†©</sup> Simon C. Roberts, <sup>∇,‡</sup> Sarah Choyke, <sup>§</sup> Karl Oetjen, <sup>‡</sup> Alan McAlees, <sup>∥</sup> Nicole Riddell, <sup>∥</sup> Robert McCrindle, <sup>⊥</sup> P. Lee Ferguson, <sup>§</sup> Christopher P. Higgins, <sup>\*,‡</sup> and Jennifer A. Field\*, <sup>#</sup>

Formula	Exact Mass	Structure
C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> -	59.0139	OH O-
CF <sub>3</sub>	68.9958	F F
NO <sub>2</sub> S <sup>-</sup>	77.9655	0==5
O₃S`	79.9574	

Barzen-Hanson, et al. (2017) DOI: 10.1021/acs.est.6b05843



Cite This: Environ. Sci. Technol. 2019, 53, 4717–4727

pubs.acs.org/est

Identification of Per- and Polyfluoroalkyl Substances in the Cape Fear River by High Resolution Mass Spectrometry and Nontargeted Screening

James McCord<sup>†®</sup> and Mark Strynar\*\*<sup>‡®</sup>





## Sources of Uncertainty or Bias in the NTA Workflow

Sample Preparation

Chromatography

Mass Spectrometry

Data Analysis

- Sample Contamination
- Selective extraction (e.g., WAX)
- Method performance
  - RT accuracy and precision
  - Interlaboratory comparability

- Method performance
  - m/z accuracy and precision
  - sensitivity
  - Interlaboratory comparability
  - MS2 spectrum uncertainty

- Unambiguous or unclear compound identity
- Non-comprehensive suspect list
- Matching to low quality library mass spectra





## NTA Study Design and Reporting

- Within the design of the study itself, there are uncertainties/biases that must be considered.
  - Does your experimental design <u>provide</u> <u>enough evidence</u> to identify a novel PFAS?
  - How has the sample collection been designed to ensure no false positives?
  - What is the performance of the method?
  - How can your results be compared to the results of other NTA studies?
  - What materials, data, and/or data tools will you use to validate your method?



Cite This: Environ. Sci. Technol. 2018, 52, 11975-11976

Viewpoint

pubs.acs.org/est

#### Is Nontargeted Screening Reproducible?

Ronald A. Hites\*®

School of Public and Environmental Affairs, Indiana University, Bloomington, Indiana 47405, United States

Karl J. Jobst\*

Department of Chemistry and Chemical Biology, McMaster University, Hamilton, Ontario L8S 4M1, Canada





## Benchmarking and Performance for Non-Targeted Analysis (BP4NTA)

- In 2018, NTA researchers formed the BP4NTA Working Group, based on the common interests and values:
  - Shared terms and definitions can lead to better comparison of methods and results.
  - We can determine benchmark performance once methods and results reporting are more harmonized.
  - Once benchmarks are set, we can then tackle concepts like "proficiency testing" for NTA measurements.

BP4NTA Introduction Manuscript: Place, et al. (2021) *Anal. Chem. In Revision*.



https://nontargetedanalysis.org



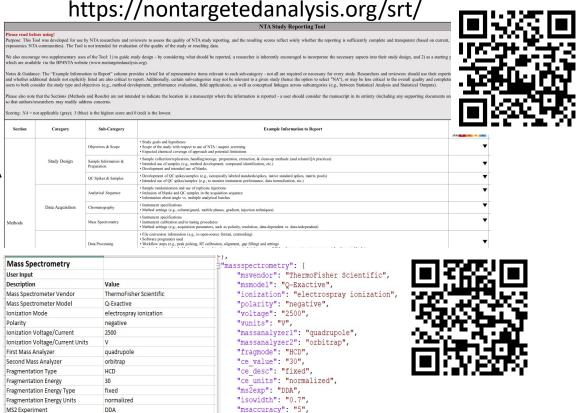


## **Method Reporting Tools**

- BP4NTA Members developed the NTA Study Reporting Tool (SRT)
  - Designed to enable study designers, manuscript reviewers, and journal editors to better understand what type of information should be considered and provided in an NTA manuscript.

Peter, et al., *Anal. Chem.* 2021. DOI: 10.1021/acs.analchem.1c02621

- To enable systematic collection of NTA method metadata for the NIST database, a macro-enable Microsoft Excel workbook has been developed.
  - Vocabulary is controlled
  - Exports data to open JSON format



"ms1resolution": "70000",
"ms2resolution": "17500",
"source": ""







## Reporting PFAS Identities

- Proving that an identified PFAS is **new** requires a significant effort to check prior lists (and study results) for all previously-discovered PFAS
  - Not all are easily searchable.
  - Naming conventions vary between researchers.
  - Compounds with the same molecular formula can have vastly different chemistries.

N-methyl-N-(3-((3,3,4,4,4-pentafluorobutyl)sulfonamido)propyl)glycine

OH Chemical Formula:  $C_{10}H_{17}F_5N_2O_4S$ Exact Mass: 356.0829

3-((*N*-(3-(dimethylamino)propyl)-1,1,2,2,2-pentafluoroethyl)sulfonamido)propanoic acid

Chemical Formula: C<sub>10</sub>H<sub>17</sub>F<sub>5</sub>N<sub>2</sub>O<sub>4</sub>S Exact Mass: 356.0829





## **NIST Suspect List of Possible PFAS**

- Purpose: To create a central location for PFAS identified by peer-reviewed manuscripts, other documentations, or computational tools.
  - Identities are attributed to source.
  - Structures (as InChI) are the key value (not name, acronym, elemental formula, etc.).
  - Index values are constant
  - Additional information (acronym, chemical class, etc.) available.
- Build upon efforts by Colorado School of Mines, Oregon State University, and others.
- Public contributions are allowed, does not require chemical registration or other official entry protocol.

## NIST Suspect List of Possible Per- and Polyfluoroalkyl Substances

				_		
ID	CHEMICAL_NAME	INCHI	SMILES	INCHIKE	FORMUL	FIXEDMASS I
:	1 1H-Perfluoro-3,3-bis(trifluorom	InChl=1S/C8HF13/c1-2-3(6(13,14)15,7(16,17)18)4(9,10)5(11,12)8(19,20)21/h1H	C#CC(C(C(	(UEYXJPFQ.	C8HF13	343.9870669
1	Bis(heneicosafluorodecyl)phosp	InChI=1S/C20H9F34O2P/c21-5(22,7(25,26)9(29,30)11(33,34)13(37,38)15(41,42)17	C(CP(=O)(	DEENCVD	C20H9F34	957.9797256
3	Tris(2-(perfluorododecyl)ethyl)	InChl=1S/C42H12F75O4P/c43-7(44,10(49,50)13(55,56)16(61,62)19(67,68)22(73,74	C(COP(=O	) JKDZBIHM	C42H12F7	2035.927562
4	2-Chloro-2-propenoic acid 3,3,4	InChI=1S/C9H6CIF9O2/c1-4(10)5(20)21-3-2-6(11,12)7(13,14)8(15,16)9(17,18)19/h	C=C(C(=O)	CVMPVWS	C9H6ClF90	351.9912611
į	Bicyclo[2.2.1]hept-2-ene, 5,5,6-	InChl=1S/C10H6F10O/c11-6(12)4-1-2-5(3-4)7(6,13)21-10(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,15)9(16,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(14,17)12(19,20)8(19,	C1=CC2CC	JZEKVGIW	C10H6F10	332.025897
(	3,3,4-Trifluoro-4-(heptafluorop	InChI=1S/C12H8F10O/c13-8(14)6-4-1-2-5(3-4)7(6)9(8,15)23-12(21,22)10(16,17)11	C1=CC2CC	NGEYGRCI	C12H8F10	358.0415471
7	5-(Nonafluorobutyl)bicyclo[2.2.	InChI=1S/C11H9F9/c12-8(13,7-4-5-1-2-6(7)3-5)9(14,15)10(16,17)11(18,19)20/h1-7	C1=CC2CC	QAIOQFZL	C11H9F9	312.0560543
8	3-(Heptafluorobutyryl)camphor	InChl=1S/C14H15F7O2/c1-10(2)6-4-5-11(10,3)8(22)7(6)9(23)12(15,16)13(17,18)14	4 CC1(C)C2C	PEWOESY	C14H15F7	348.0960273
9	Perfluorobutylsulfonamide	InChI=1S/C4H2F9NO2S/c5-1(6,3(9,10)11)2(7,8)4(12,13)17(14,15)16/h(H2,14,15,16)	C(C(C(F)(F	) FUVKFLJW	C4H2F9NC	298.9662533
		InChl=1S/C10H4Cl2F7NO/c11-5-2-1-4(3-6(5)12)20-7(21)8(13,14)9(15,16)10(17,18)				356.9558166
		InChl=1S/C10H4Cl2F7NO/c11-4-1-5(12)3-6(2-4)20-7(21)8(13,14)9(15,16)10(17,18)				
12	2 2-Propenoic acid, 2-methyl-, 2-	InChl=1S/C28H28F17N3O8S/c1-6-48(9-10-54-19(50)46-16-8-7-14(4)17(11-16)47-2	CCN(CCOC	WUGSPVX	C28H28F1	889.1325656

https://github.com/USNISTGOV/NISTPFAS





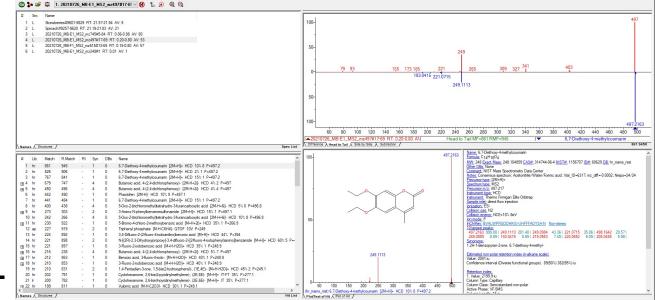


## Mass Spectral Libraries

- For suspect screening analysis, libraries containing mass spectra produced from chemical standards can be extremely powerful.
- Mass spectral matches provide some of the strongest evidence for identification of compounds in lieu of analytical standards.
- Conventionally, most mass spectral libraries contain spectra for compounds that have commerciallyavailable standards

NIST MS Search Program https://chemdata.nist.gov









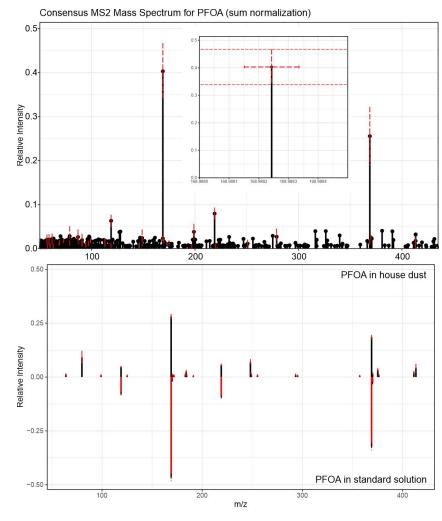
## **Uncertainty of Mass Spectra**

- 2021 published study:
  - There is quantifiable variability about the measured m/z and intensity for fragmentation ions.
  - Sample matrix may affect the presence and/or variability of fragment ions for the same compound.
  - By calculating the variability of fragmentation mass spectra, the confidence of mass spectral match can be assessed.

Place, BJ, JASMS (2021)

DOI: 10.1021/jasms.0c00423



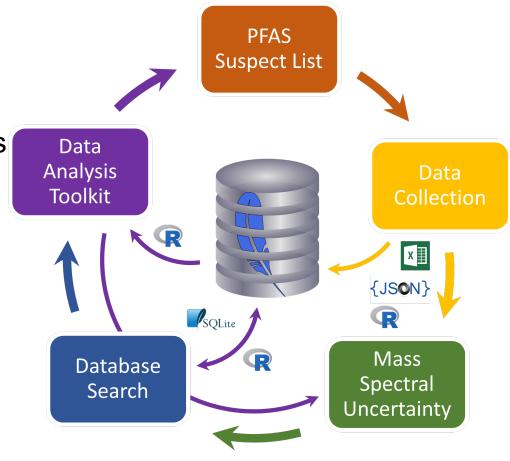




## **NIST PFAS MS Database**

## **ER20-1056 Project:**

- Creation of a database infrastructure to include:
  - Structure-specific PFAS suspects
  - Reference mass spectra for identified PFAS
  - Mass spectra of PFAS in standards and matrix samples with the variability assessed
  - Open schema (SQLite) that enables researchers to build advanced tools for data analysis



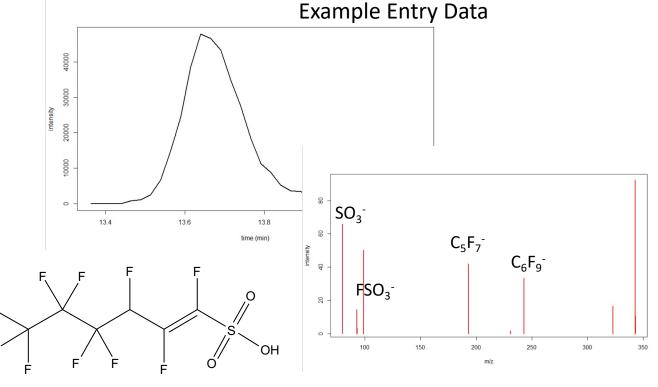




### **NIST PFAS MS Database**

## **ER20-1056 Project:**

- NIST PFAS Suspect List
  - 4,884 individual structures
- Current database population
  - Around 291 mass spectra from standards and commercial mixtures.
  - Representing 99 individual PFAS<sub>F</sub>
  - Multiple energy levels, different instrumentation.
- Open to submission!



(Z)-1,2,3,4,4,5,5,6,6,6-decafluorohex-1-ene-1-sulfonic acid

Chemical Formula: C<sub>6</sub>H<sub>2</sub>F<sub>10</sub>O<sub>3</sub>S

Exact Mass: 343.9565

Source: Barzen-Hanson et al. 2017





## Additional Knowledge Gaps

- In lieu of reference mass spectra, in silico fragmentation prediction can be a powerful tool, but there is a lack of understanding of the accuracy of these predictions for PFAS.
- Benchmark performance for any NTA workflow needs to be established.
  - Requires common communication of performances in study results
  - "Good" performance must be empirically determined.
- There are no reference materials or reference data for evaluating and/or comparing NTA performance between laboratories.





### Conclusions

- Non-targeted analysis and suspect screening analysis are powerful tools for the identification of unknown PFAS.
  - But there are sources of uncertainty and bias that must be addressed within any NTA workflow.
- Through the SERDP project, NIST is developing reference data and data tools to address current NTA data gaps.
- Future webinars and training programs are being developed.
- An interlaboratory study (and first glance at using the new database!) is planned for the end of 2022.
- If you would be interested in a training and/or the interlaboratory study please contact us at <a href="mailto:pfas@nist.gov">pfas@nist.gov</a>





## Acknowledgements

- This project has been supported by SERDP, project ER20-1056
- NIST PFAS Researchers:
  - Jared Ragland, Nik Blonder, Alix Rodowa, Kathy Peter, Jessica Reiner
- Collaborators:
  - Field Group (OSU), Higgins Group (CSM), Xiao Group (UND), Helbling Group (Cornell), Ferguson Group (Duke)

